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CHEMISTRY 9701/01 1 Hour

10 When Vanadium(II) Compounds Are Dissolved In Water, The Following Equilibrium Is Established. $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O + OH^-$ What Would Alter The Composition Of The Equilibrium Mixture In Favour Of The V^{2+} Ions? A Adding An Acid B Adding A Reagent That Selectively Precipitates Mar 13th, 2022

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Chemistry: Atomic Structure Page 3 Question 2 9701_w/14/qp23 May 3th, 2022

CHEMISTRY 9701/04

(c) (i) Write A Chemical Equation Representing The Lattice Energy Of AgBr. (ii) Use The Following Data To Calculate A Value For The Lattice Energy Of AgBr(s). First Ionisation Energy Of Silver = +731kJmol⁻¹ Electron Affinity Of Bromine = 325kJmol⁻¹ Enthalpy Change Of Atomisation Of Silver = +285kJmol⁻¹ Enthalpy Change Of Atomisation Of Bromine = +112kJmol⁻¹ ... Jan 22th, 2022

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Mark Required For Grade: Maximum Raw Mark ... S6 12, 62 Feb 9th, 2022

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